



**Question (1)**

**(18 Marks)**

- 1.1 Environmental pollution problems must be faced globally. Why? (2)
- 1.2 Define the following terms:
  - a) Ecology
  - b) Pollution
- 1.3 Explain with sketch the characteristics and position of ionosphere layer? (3)
- 1.4 What are the benefits of ozone? (2)
- 1.5 What are the main reasons of global warming phenomena? (2)
- 1.6 Explain how can wind energy help in environment conservation? (3)
- 1.7 Calculate the BOD<sub>s</sub>, given the following data:
 

Temperature of sample, 20 °C. Initial DO is 9.2 mg/l. Dilution is 1:30, with seeded dilution water. Final DO of seeded dilution water is 8 mg/l. Final DO bottle with sample and seeded dilution water is 2 mg/l. Volume of BOD bottle is 300 ml. If the temperature of the sample is increased to 30 °C, would the BOD<sub>s</sub> greater or smaller and why? (4)

**Question (2)**

**(18 Marks)**

- 2.1 Explain how heat is considered a source of water pollution? (2)
- 2.2 You are engineer working in a sewer and water company in a certain city. The weather in this city is dry. It is required to build a new sewer system in this city. Would you prefer the design of separate sewer or combined sewer and why? (3)
- 2.3 The concentration of Calcium in water is 50 mg/l. What is the concentration of calcium in this water in ppm? (2)
- 2.4 What is the main disadvantage of Winkler Dissolved Oxygen test? (2)
- 2.5 When are the dilution and seeding of water sample used? (2)
- 2.6 How can bacteriological test of water quality be done? (3)
- 2.7 The following data were obtained for a sample:
 

Total solids = 4000 mg/l  
 Suspended solids = 5000 mg/l  
 Volatile suspended solids = 2000 mg/l  
 Fixed suspended solids = 1000 mg/l.  
 Which of these numbers is questionable and why? (4)

**Question (3)**

**(18 Marks)**

- 3.1 In water treatment field, what is *coagulation and flocculation*? (2)
- 3.2 What are the main common sense rules that must be considered for routing trucks optimization? (2)
- 3.3 How can particle size be controlled in vertical hammer mill in solid waste shredding? (2)
- 3.4 The first step in solid waste separation operation is the decrease of solid waste size. Why? (2)
- 3.5 Describe with sketch the function of Magnets as a solid waste separator? (3)
- 3.6 Describe in detail with sketch the design and construction of hazardous waste landfill? (3)
- 3.7 You are asked to design a resource recovery (materials separation) system for the following waste:

Component	Fraction by Weight
Newspaper	70
Glass bottles	15
Steel cans	10
Aluminum cans	0
Plastics	5
Garbage	0

- a) Design such a system and draw a schematic diagram.
- b) Discuss the next steps applied on the separated materials to make use of their components (4)

**Question (4)**

**(18 Marks)**

- 4.1 Distinguish between infrasound and ultrasound? (2)
- 4.2 Define white noise? (2)
- 4.3 In acoustic science what is meant by  $L_{10} = 70$  dB(A)? (3)
- 4.4 What is the main difference between primary air pollutant and secondary air pollutant? (2)
- 4.5 What is synergistic? Give two examples of synergism in air pollution? (2)
- 4.6 Which of the following are more dangerous on respiratory system and why?

Soluble or no soluble particulates

(2)

4.7 Draw a map with X and Y coordinates (X horizontal, Y vertical) and place on the map the following:

- Industrial Plant "A" at X = 3, Y = 3
- Industrial Plant "B" at X = 8, Y = 1
- Industrial Plant "C" at X = 8, Y = 8
- Air Sampling Station at X = 5, Y = 5

The data at the air sampling station are:

Day	1	2	3	4	5	6	7	8	9	10
Wind Direction	N	NE	NW	N	NE	SW	S	SW	E	W
Particulates ( $\mu\text{g}/\text{m}^3$ )	80	120	30	90	130	20	30	40	100	10
SO <sub>2</sub> ( $\mu\text{g}/\text{m}^3$ )	80	20	30	40	20	180	100	200	60	100

Draw pollution roses to show which plant is guilty of the air pollution.

(5)

**Question (5)**

**(18 Marks)**

- 5.1 Specify with sketch the characteristics of both high pressure and low pressure regions? Which is preferable for better air quality condition? (2)
- 5.2 Is the emission of pollutants during the night in a certain city more dangerous or the emission during the day and why? (2)
- 5.3 Show with example and sketch that super-adiabatic lapse rate is better for the dispersion of pollutants than sub-adiabatic lapse rate. (3)
- 5.4 How can respirable particulate be measured? (2)
- 5.5 How can ozone concentration in a sample of air be measured? (2)
- 5.6 Describe with sketch the operation of bag filter for controlling of particulates? (3)
- 5.7 Given the following temperature soundings:

Elevation (m)	0	50	100	150	200	250	300
Temperature (°C)	20	15	10	15	20	15	20

What type of plume would you expect if the exit temperature of the plume were 15°C and the smoke stack were: a) 50 m tall? b) 50 m tall? c) 250 m tall? (4)

With best wishes

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This exam measures the following ILOs											
Question Number	Q1-1,1-2,2-4,4-2	Q1-3,2-6,3-2,5-7,4-5	Q1-4,2-2,4-6,5-4	Q1-5,2-3,3-1	Q1-6,1-5,5-3	Q2-3,3-5,2-7	Q1-4,2-5,3-2,4-7,5-2	Q1-6,1-7,2-1,3-4,5-5	Q1-4,2-2,3-1,3-6,4,4,5-1	Q1-6,1-5,5-3	Q1-6,1-5,5-3
Skills	A6	A11	A12	A13	A18	B5	B9	C3	C15	D1	D2
	Knowledge & Understanding Skills					Intellectual Skills		Professional Skills		General Skills	